

Synthesis and receptor binding properties of 2 β -alkynyl and 2 β -(1,2,3-triazol)substituted 3 β -(substituted phenyl)tropane derivatives

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Abstract—A series of 2 β -alkynyl and 2 β -(1,2,3-triazol)substituted 3 β -(substituted phenyl)tropanes were synthesized and evaluated for affinities at dopamine, serotonin, and norepinephrine membrane transporters using competitive radioligand binding assays. All tested compounds were found to exhibit nanomolar or subnanomolar affinity for the dopamine transporter (DAT). One of the most potent and selective compounds in the series was 3 β -(4-chlorophenyl)-2 β -(4-nitrophenylethynyl)tropane (**10c**) that possessed an IC₅₀ value of 0.9 nM at the DAT and K_i values of 230 nM and 620 nM at the norepinephrine transporter (NET) and serotonin transporter (5-HTT), respectively.

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1. Introduction

Cocaine (**1**, Fig. 1) is one of the most powerful central nervous system (CNS) stimulants with reinforcing properties. While cocaine blocks the presynaptic reuptake of dopamine (DA), serotonin (5-HT), and norepinephrine (NE), numerous studies have strongly supported the hypothesis that the dopamine transporter (DAT) is highly significant in cocaine abuse regarding its reinforcing effects.^{1–5} Although other targets may be important to addiction (metabotropic and ionotropic glutamate receptors, GABA)^{6–8} animal behavioral and clinical finding suggest that the DAT is still an important target.^{9–14} Structure–activity relationship (SAR) studies of a number of different classes of DAT inhibitors with the goal of the development of pharmacotherapies to treat cocaine addiction have been reported.^{12,15–18} One of the most studied classes of DAT inhibitors are the 3-phenyltropanes.^{12,15,19,20} The lead compound was 3 β -phenyltropane-2 β -carboxylic acid methyl ester (**2a**, WIN 35,065-2).^{21,22} A large part of our SAR studies have been directed toward modification of the 4'-methyl and 4'-chloro analogs **2b,c**.¹²

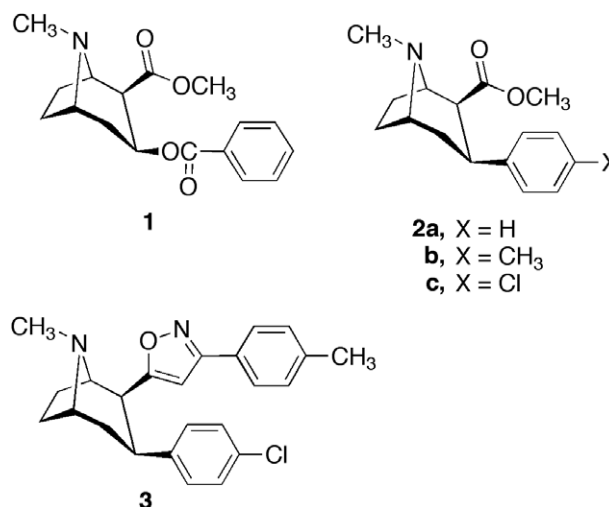


Figure 1. Structures of compounds **1**, **2a–c**, and **3**.

Most of the studies were directed toward changes in the C(2)-position of **2a–c**. SAR studies have revealed that a variety of functional groups and substituents are well tolerated at this position without loss of high-affinity for monoamine transporters. In early studies, we showed that a variety of 2 β -esters and amides had high-affinity for the DAT, in some cases, with considerably reduced affinity at the 5-HTT and NET.¹² SAR

Keywords: Monoamine transporters; 3-Phenyltropanes; Alkynes; 1,2,3-Triazoles; Cocaine; Addiction.

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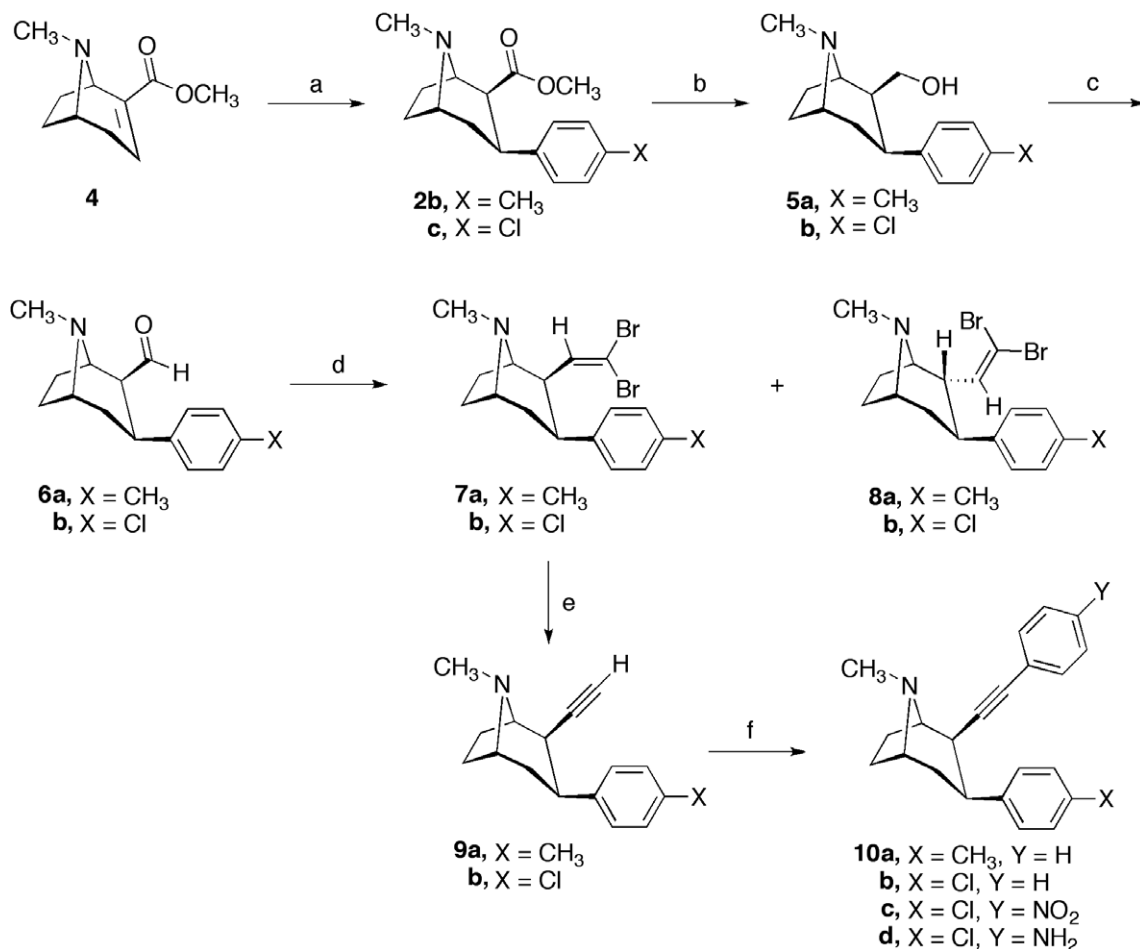
studies from other groups and us also revealed that large lipophilic groups on the 2 β -position, including alkyl, alkenyl and aryl substituents retained high DAT binding affinity.^{12,23–25} In addition, exchange of the 2 β -carbomethoxy group with bioisosteric heterocyclic groups led to analogs with high-affinity and selectivity for the DAT. One of the most studied compounds in this series is the DAT selective 3-phenyltropane analog, RTI-336 (**3**). RTI-336 is currently in advanced preclinical development.^{10–12}

Despite extensive efforts directed toward the development of a pharmacotherapy for cocaine abuse, at present no clinically approved drugs are available. In order to gain a better understanding of the molecular mechanisms of cocaine actions in the brain and find highly potent and selective monoamine uptake inhibitors, we have continued the investigation on modification of **2a**. The present study was undertaken to further explore the SAR of 2-substituents of the 3 β -phenyltropanes. In this paper, we describe the synthesis and monoamine transporter binding properties of several 2 β -alkynyl and 2 β -(1,2,3-triazol)substituted 3 β -(substituted phenyl)tropane derivatives, and report that 3 β -(4-chlorophenyl)-2 β -(4-nitrophenylethynyl)tropane (**10c**)

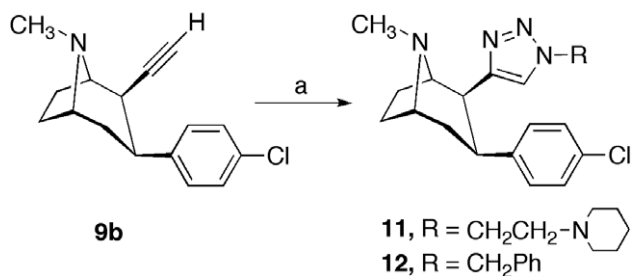
has high potency and good selectivity for the DAT relative to the 5-HTT and NET.

2. Chemistry

The synthesis of 2 β -alkynyl-3 β -(substituted phenyl)tropane analogs **9a,b** and **10a–d** starting with anhydroecgonine methyl ester (**4**) is outlined in Scheme 1. The 1,4-addition of **4** with the appropriate Grignard reagent at -45°C in ethyl ether followed by trifluoroacetic acid (TFA) led to the corresponding 3 β -aryl substituted compounds **2b,c**.²⁶ Lithium aluminum hydride (LiAlH_4) reduction of the 2 β -ester group of **2b,c** afforded alcohols **5a,b**. Swern oxidation of **5a,b** provided aldehydes **6a,b**. The aldehydes **6a,b** were not stable and underwent epimerization at the C(2)-position during silica gel chromatography. Therefore, crude **6a,b**, used without purification, were treated with carbon tetrabromide, triphenylphosphine, and zinc to give the 2 β -isomers **7a,b** in the order of 54–56% yield as well as the minor 2 α -isomers **8a,b** in 1.7% and 1.4% yield, respectively, after column chromatography. With the C(2)-position no longer susceptible to epimerization, **7a,b** were treated with 2 equivalents of butyl lithium to afford 2 β -ethynyltro-



Scheme 1. Reagents and conditions: (a) Grignard reagent, -45°C , 2 h, then -78°C , TFA; (b) LiAlH_4 , THF, room temperature; (c) Swern oxidation; (d) CBr_4 , PPh_3 , Zn , CH_2Cl_2 , room temperature; (e) BuLi , THF, -78°C to room temperature; (f) ArI , $\text{Pd(PPh}_3)_4$, CuI , benzene– Et_3N , 40°C .



Scheme 2. Reagents and conditions: (a) azide, CuSO₄, sodium ascorbate, *t*BuOH-H₂O, room temperature.

panes **9a,b** exclusively. Sonogashira coupling of ethyne **9a** or **9b** with iodobenzene using tetrakis(triphenylphosphine)palladium(0), and copper(I) iodide in 1:1 mixture of benzene-triethylamine gave 2 β -phenylethynyltropanes **10a** and **10b**, respectively. Finally, coupling of **9b** with 1-iodo-4-nitrobenzene or 4-iodoaniline furnished **10c** and **10d**. The relative stereochemistry of each compound was determined by ¹H NMR spectral analysis, particularly with the aid of coupling constants of C(2)–H and C(3)–H. The vicinal couplings of $J_{2eq,3ax} = 5.4\text{--}5.7$ Hz and $J_{2ax,3ax} = 11.5\text{--}11.9$ Hz for the 2 β - and 2 α -substituents, respectively, are in good agreement with stereochemical assignments.

Recently, the copper(I)-catalyzed 1,2,3-triazole formation from terminal alkynes and azides, also known as the ‘click reaction’, has found growing applications in drug discovery due to the favorable physicochemical properties of triazole, which can readily associate with biological targets through hydrogen bonding and dipole interactions.^{27,28} Accordingly, 2 β -(1,2,3-triazol)substituted tropanes **11** and **12** were synthesized by treatment of **9b** with 1-(2-azidoethyl)piperidine or benzylazide in the presence of copper sulfate and sodium ascorbate (**Scheme 2**).

3. Biology

The binding affinities for the target compounds at the DAT, NET, and 5-HTT were determined via competitive binding assays using the previously reported procedures.^{29,30} The final concentration of radioligands in the assays was 0.5 nM [³H]WIN35,428 for the DAT, 0.5 nM [³H]nisoxetine for the NET, and 0.2 nM [³H]paroxetine for the 5-HTT. The results of the binding studies, along with binding data of cocaine and **2a**¹⁰ for comparison are listed in **Table 1**. Since the NET and 5-HTT have only one binding site, K_i values were calculated for inhibition of binding at these two transporters.

Table 1. Monoamine transporter binding potencies for 2 β -substituted 3 β -(substituted phenyl)tropane derivatives

Compound ^a	<i>c</i> log <i>P</i>	Structure				DAT, IC ₅₀ ^d (nM) [³ H]WIN35,428	NET, K_i ^d (nM) [³ H]nisoxetine	5-HTT, K_i ^d (nM) [³ H]paroxetine
			X	Y	R			
Cocaine ^b	—		—	—	—	89.1	1990	95
2a	—	—	—	—	—	23	556	182
7a	5.21	A	CH ₃	—	—	0.23 ± 0.05	3.4 ± 0.6	39 ± 10
7b	5.35	A	Cl	—	—	0.32 ± 0.05	3.5 ± 0.3	7 ± 2
8a	5.21	A ^c	CH ₃	—	—	3.6 ± 1.1	55 ± 2	252 ± 13
8b	5.35	A ^c	Cl	—	—	1.4 ± 0.2	49 ± 2	76 ± 9
9a	3.40	B	CH ₃	—	—	11 ± 3	150 ± 20	54 ± 15
9b	3.53	B	Cl	—	—	3.6 ± 0.4	75 ± 13	18 ± 2
10a	5.87	C	CH ₃	H	—	0.8 ± 0.3	31 ± 1	730 ± 190
10b	6.01	C	Cl	H	—	0.8 ± 0.2	14 ± 2	81 ± 4
10c	5.74	C	Cl	NO ₂	—	0.9 ± 0.2	230 ± 20	620 ± 150
10d	4.73	C	Cl	NH ₂	—	2.2 ± 0.5	50 ± 3	19 ± 3
11	3.89	D	Cl	—	NCH ₂ CH ₂	8 ± 1	430 ± 50	138 ± 11
12	4.25	D	Cl	—	C ₆ H ₅ CH ₂	47 ± 12	2560 ± 130	161 ± 27

^a All compounds were tested as the HCl salt.

^b Data taken from Ref. 9.

^c 2 α -Stereoisomer.

^d All values are means ± standard error of three or four experiments performed in triplicate.

4. Results and discussion

One of the most interesting features of the SAR of 3-phenyltropane analogs for the DAT is that exchanging the 2 β -ester group in WIN 35,065-2, with amide, ether, heterocyclic, keto, alkyl, alkenyl, or aryl substituents at the C(2)-position provides compounds with equal or greater affinity than that of WIN 35,065-2 at the DAT.¹² Although the DAT tolerates ligands having a broad variety of 2 β -substituents with little change in the affinity of the ligand, the nature of the substituents has a profound effect on the monoamine selectivity. In order to gain additional information on the structural requirements for high-affinity and good selectivity at the C(2)-position, we designed and synthesized a new series of 2-substituted 3 β -(substituted phenyl)tropane derivatives.

All the compounds possessed high affinities at the DAT with nanomolar or subnanomolar IC₅₀ values ranging from 0.23 nM to 47 nM. No correlation was observed between the DAT binding affinities with the calculated partition coefficients (*clogP*) of the compounds. Generally, the 3 β -(4-chlorophenyl) substituted tropanes except **7b** were equal or more potent than the compounds with a 3 β -(4-methylphenyl) substitution. The 2 α -isomers **8a** and **8b** were approximately 16- and 4-fold less potent than the corresponding 2 β -isomers **7a** and **7b**, respectively. This was consistent with the previous findings that exchange of the 2 β -substituent in the tropane derivatives with the 2 α moiety decreased the relative binding affinity for the DAT.³¹

Among the tested 2 β -substituted tropanes, the 2 β -dibromovinyltropanes **7a** and **7b** possessed the highest potency at the DAT with IC₅₀ values of 0.23 nM and 0.32 nM, respectively. Replacement of the vinyl group with the ethynyl substitution afforded **9a** and **9b** with decreased affinities at the DAT (11 nM vs 0.23 nM and 3.6 nM vs 0.32 nM IC₅₀). It is interesting to note that attachment of a phenyl group to the end of the ethyne group gave **10a** and **10b**, which regained the binding affinity by 4- to 14-fold, respectively. In addition, replacing the phenylethynyl group at the C(2)-position of the 3 β -(4-chlorophenyl) analogue **10b** (IC₅₀ = 0.8 nM) with the 4-nitrophenylethynyl substitution to give **10c** had no effect on binding affinity (IC₅₀ = 0.9 nM). However, the addition of a 4-amino group to **10b** resulted in approximately 3-fold loss of binding affinity at the DAT. These findings support previous reports that a large pocket is present in the DAT binding site occupied by the 2 β -substituent.¹² The lipophilic interactions as well as the possible π -interactions between the 2 β -substituents and the binding site may play an important role for the high affinity of these ligands. Finally, as analogs of the reported 2 β -heterocyclic tropanes, the 2 β -(1,2,3-triazol)substituted tropane derivatives **11** and **12** also possessed high-affinity at the DAT with IC₅₀ values of 8 nM and 47 nM, respectively.

In terms of monoamine selectivity, all the compounds exhibited lower binding affinities at the NET and 5-HTT relative to the DAT. However, no correlation

was observed between the 2 β -substituents and the monoamine selectivity. 3 β -(4-Chlorophenyl)-2 β -(4-nitrophenylethynyl)tropane (**10c**) with an IC₅₀ value of 0.9 nM at the DAT and K_i values of 230 nM at the NET and 620 nM at the 5-HTT, respectively, was one of the most potent and selective compounds for the DAT relative to the NET and 5-HTT in the series.

In summary, several novel 3 β -(substituted phenyl)tropanes with various 2 β -alkynyl and 2 β -(1,2,3-triazol) substituents were synthesized and evaluated for their monoamine transporter binding affinities. All the tested compounds demonstrated higher potency at the DAT than cocaine and were more selective relative to binding at the NET and 5-HTT. One of the most potent and selective compounds in the series, 3 β -(4-chlorophenyl)-2 β -(4-nitrophenylethynyl)tropane (**10c**) had an IC₅₀ value of 0.9 nM at the DAT and was 256- and 689-fold selective for the DAT over the NET and 5-HTT, respectively. These 2 β -substituted tropanes are promising leads for further investigation of highly potent and selective monoamine inhibitors.

5. Experimental

Melting points were determined using a MEL-TEMP II capillary melting point apparatus and are uncorrected. Nuclear magnetic resonance (¹H NMR and ¹³C NMR) spectra were obtained on a Bruker Avance DPX-300 MHz NMR spectrometer. Chemical shifts are reported in parts per million (ppm) with reference to internal solvent. Mass spectra (MS) were run on a Perkin-Elmer SCIEX API 150 EX mass spectrometer outfitted with ESI (turbospray) source or on a Hewlett Packard 5989A instrument by electron impact. Elemental analysis was performed by Atlantic Microlab Inc., Atlanta, GA. Optical rotations were measured on an AutoPol III polarimeter, purchased from Rudolf Research. Analytical thin-layer chromatography (TLC) was carried out using EMD silica gel 60 F₂₅₄ TLC plates. TLC visualization was achieved with a UV lamp or in an iodine chamber. Flash column chromatography was done on a CombiFlash Companion system using Isco prepacked silica gel columns or using EM Science silica gel 60 Å (230–400 mesh). Unless otherwise stated, reagent-grade chemicals were obtained from commercial sources and were used without further purification. All moisture- and air-sensitive reactions and reagent transfers were carried out under dry nitrogen.

5.1. 3 β -(4-Methylphenyl)-2 β -hydroxymethyltropane (**5a**)

To a stirred solution of **2b**²⁶ (11.4 g, 0.04 mol) in anhydrous THF (100 mL) at 0 °C under nitrogen was added LiAlH₄ (3.04 g, 0.080 mol). After stirring at room temperature for 2 h, the reaction was quenched by slow addition of H₂O (20 mL). The organic layer was separated and the aqueous layer was extracted with EtOAc (3 × 30 mL). The combined organic phases were washed with brine (3 × 50 mL), and dried (Na₂SO₄). Removal of the solvent under reduced pressure afforded crude **5a** (9.60 g, 98%) as a white solid; mp 78–80 °C; ¹H NMR

(CDCl₃) δ 7.25 (d, J = 8.1 Hz, 2 H), 7.12 (d, J = 8.1 Hz, 2 H), 3.75 (dd, J = 10.8, 2.1 Hz, 1H), 3.51–3.45 (m, 1H), 3.39 (dd, J = 10.8, 2.1 Hz, 1H), 3.37–3.29 (m, 1H), 3.05 (ddd, J = 12.9, 5.7, 5.7 Hz, 1H), 2.51 (ddd, J = 12.9, 12.9, 3.0 Hz, 1H), 2.32 (s, 3H), 2.27 (s, 3H), 2.26–2.04 (m, 2H), 1.82–1.57 (m, 3H), 1.52–1.43 (m, 1H); MS (ESI) m/z 246.4 (M+1). The desired compound was used in the next step without further purification.

5.2. 3 β -(4-Chlorophenyl)-2 β -hydroxymethyltropane (**5b**)

The procedure for **5a** was followed using 10.0 g (0.034 mol) of **2c**²⁶ to give 8.90 g (99%) of **5b** as a white solid: mp 82–84 °C; ¹H NMR (CDCl₃) δ 7.38–7.25 (m, 4H), 3.75 (dd, J = 11.1, 2.1 Hz, 1H), 3.50–3.42 (m, 1H), 3.38–3.28 (m, 2H), 3.12–3.00 (m, 1H), 2.49 (ddd, J = 12.9, 12.9, 3.2 Hz, 1H), 2.27 (s, 3H), 2.26–2.04 (m, 2H), 1.80–1.57 (m, 3H), 1.50–1.44 (m, 1H); MS (ESI) m/z 266.3 (M+1). The desired compound was used in the next step without further purification.

5.3. 3 β -(4-Methylphenyl)-2 β -(2,2-dibromovinyl)tropane (**7a**) and 3 β -(4-methylphenyl)-2 α -(2,2-dibromovinyl)tropane (**8a**)

To a stirred solution of oxalyl chloride (2 M solution, 16.5 mL, 33.0 mmol) in anhydrous CH₂Cl₂ (100 mL) at –78 °C under nitrogen was added anhydrous DMSO (4.68 mL, 66.0 mmol). After stirring for 15 min, a solution of **5a** (5.40 g, 22.0 mmol) in anhydrous CH₂Cl₂ (100 mL) was added and the reaction mixture was stirred at –78 °C for another 30 min. TEA (18.4 mL, 132 mmol) was then added and the reaction mixture was warmed to room temperature and stirred for 2 h. The reaction was quenched by addition of H₂O (10 mL). The organic layer was separated and washed with NH₄Cl (3 \times 50 mL), brine (50 mL) and dried (Na₂SO₄). Removal of the solvent under reduced pressure afforded aldehyde **6a** (5.42 g) as an oil, which was used in the next step without further purification.

To a stirred solution of CBr₄ (14.6 g, 0.044 mol) in anhydrous CH₂Cl₂ (165 mL) at 0 °C under nitrogen was added PPh₃ (11.5 g, 0.044 mol) followed by zinc dust (2.88 g, 0.044 mol). After stirring at room temperature for 16 h, the reaction mixture was cooled to 0 °C and a solution of aldehyde **6a** (5.42 g) was added. The reaction mixture was stirred at room temperature for 1 h and filtered through a short pad of Celite. The filtrate was washed with brine (3 \times 50 mL), dried (Na₂SO₄), and concentrated under reduced pressure. Flash column chromatography on silica gel (300 g) using 0 \rightarrow 30% ether in hexanes with 5% TEA afforded 2 β -isomer **7a** (4.94 g, 56%) and 2 α -isomer **8a** (0.15 g, 1.7%). Compound **7a**: white solid; mp 48–50 °C; ¹H NMR (CDCl₃) δ 7.08 (d, J = 8.1 Hz, 2H), 7.01 (d, J = 8.1 Hz, 2H), 6.70 (d, J = 9.4 Hz, 1H), 3.35–3.26 (m, 1H), 2.19–2.13 (m, 1H), 3.07 (ddd, J = 12.6, 5.7, 5.7 Hz, 1H), 2.64 (ddd, J = 5.7, 3.5, 9.4 Hz, 1H), 2.31 (s, 3H), 2.22 (s, 3H), 2.21–2.01 (m, 3H), 1.80–1.58 (m, 3H); ¹³C NMR (75 MHz; CDCl₃) δ 139.8, 138.9, 135.8, 129.0, 127.8, 87.7, 66.1, 62.4, 51.5, 42.2, 35.9, 35.1, 26.6, 25.2, 21.2; MS (ESI) m/z 400.2 (M+1). The free base was converted

to the hydrochloride salt: mp 241 °C (dec); [α]_D²⁰ –41.67° (c 0.24, CH₃OH); Anal. Calcd for C₁₇H₂₁Br₂N·H·Cl·0.75H₂O: C, 45.46; H, 5.27; N, 3.12. Found: C, 45.32; H, 5.35; N, 3.01.

Compound **8a**: white solid; mp 115–117 °C; ¹H NMR (CDCl₃) δ 7.07 (s, 4H), 6.09 (d, J = 9.1 Hz, 1H), 3.30–3.16 (m, 2H), 3.03 (ddd, J = 11.5, 2.6, 9.1 Hz, 1H), 2.60 (ddd, J = 11.5, 11.7, 5.4 Hz, 1H), 2.39 (s, 3H), 2.30 (s, 3H), 2.20–1.94 (m, 2H), 1.87 (ddd, J = 11.7, 11.7, 2.4 Hz, 1H), 1.77–1.54 (m, 3H); ¹³C NMR (75 MHz; CDCl₃) δ 140.1, 140.0, 136.2, 129.4, 127.7, 89.9, 64.4, 61.4, 49.7, 40.4, 39.9, 39.7, 26.4, 23.7, 21.2; MS (ESI) m/z 400.3 (M+1). The free base was converted to the hydrochloride salt: mp 210 °C (dec); [α]_D²⁰ +23.53° (c 0.26, CH₃OH); Anal. Calcd for C₁₇H₂₁Br₂N·H·Cl·0.25H₂O: C, 46.39; H, 5.15; N, 3.18. Found: C, 46.21; H, 5.36; N, 3.11.

5.4. 3 β -(4-Chlorophenyl)-2 β -(2,2-dibromovinyl)tropane (**7b**) and 3 β -(4-chlorophenyl)-2 α -(2,2-dibromovinyl)tropane (**8b**)

The procedure for **7a** and **8a** was followed using 5.30 g (0.02 mol) of **5b** to give 4.53 g (54%) of 2 β -isomer **7b** and 0.12 g (1.4%) of 2 α -isomer **8b**. Compound **7b**: white solid; mp 88–90 °C; ¹H NMR (CDCl₃) δ 7.30–7.20 (m, 2H), 7.10–7.03 (m, 2H), 6.70 (d, J = 9.5 Hz, 1H), 3.32–3.26 (m, 1H), 3.18–3.10 (m, 1H), 3.07 (ddd, J = 12.9, 5.6, 5.4 Hz, 1H), 2.62 (ddd, J = 5.6, 3.3, 9.5 Hz, 1H), 2.22 (s, 3H), 2.20–2.02 (m, 3H), 1.80–1.57 (m, 3H); ¹³C NMR (75 MHz; CDCl₃) δ 140.5, 139.2, 132.0, 129.3, 128.2, 88.2, 65.9, 62.1, 51.3, 42.1, 36.0, 35.0, 26.6, 25.2; MS (ESI) m/z 420.3 (M+1). The free base was converted to the hydrochloride salt: mp 235 °C (dec); [α]_D²⁰ –34.9° (c 0.22, CH₃OH); Anal. Calcd for C₁₆H₁₈Br₂ClN·HCl: C, 42.14; H, 4.20; N, 3.07. Found: C, 42.27; H, 4.21; N, 3.14.

Compound **8b**: oil; ¹H NMR (CDCl₃) δ 7.25 (d, J = 8.4 Hz, 2H), 7.13 (d, J = 8.4 Hz, 2H), 6.07 (d, J = 9.4 Hz, 1H), 3.30–3.16 (m, 2H), 3.00 (ddd, J = 11.9, 2.4, 9.4 Hz, 1H), 2.61 (ddd, J = 11.9, 12.2, 5.4 Hz, 1H), 2.39 (s, 3H), 2.20–1.94 (m, 2H), 1.85 (ddd, J = 12.2, 12.6, 2.4 Hz, 1H), 1.77–1.53 (m, 3H); ¹³C NMR (75 MHz; CDCl₃) δ 141.7, 139.4, 132.4, 129.3, 128.8, 90.4, 64.5, 61.4, 50.1, 40.3, 40.1, 39.6, 26.3, 23.7; MS (ESI) m/z 420.5 (M+1). The free base was converted to the hydrochloride salt: mp 140 °C (fusion); [α]_D²⁰ +26.1° (c 0.23, CH₃OH); Anal. Calcd for C₁₆H₁₈Br₂ClN·HCl: C, 42.14; H, 4.20; N, 3.07. Found: C, 42.27; H, 3.97; N, 3.02.

5.5. 3 β -(4-Methylphenyl)-2 β -ethynyltropane (**9a**)

To a stirred solution of **7a** (400 mg, 1.00 mmol) in anhydrous THF (10 mL) at –78 °C under nitrogen was added BuLi (1.6 M solution, 1.31 mL, 2.10 mmol). After stirring at –78 °C for 1 h, the reaction mixture was warmed to room temperature and the stirring was continued for another 1 h. The reaction mixture was quenched by addition of saturated NH₄Cl. The organic layer was separated and the aqueous layer was extracted

with EtOAc (3 × 30 mL). The combined organic phases were washed with brine (3 × 30 mL), dried (Na₂SO₄), and concentrated under reduced pressure. Flash column chromatography on silica gel (12 g Isco prepacked column) using 0 → 5% ether in hexanes with 5% TEA afforded **9a** (130 mg, 54%) as an oil: ¹H NMR (CDCl₃) δ 7.22–7.08 (m, 4H), 3.38–3.28 (m, 2H), 2.97 (ddd, *J* = 12.9, 5.4, 5.4 Hz, 1H), 2.73–2.67 (m, 1H), 2.40–2.00 (m, 10H), 1.76–1.51 (m, 3H); ¹³C NMR (75 MHz; CDCl₃) δ 139.4, 136.0, 128.8, 128.1, 84.8, 71.6, 66.7, 61.9, 42.3, 41.6, 35.9, 35.1, 26.1, 25.2, 21.2; MS (ESI) *m/z* 240.4 (M+1). The free base was converted to the hydrochloride salt: mp 210 °C (dec); [α]_D²⁰ –114.2° (*c* 0.28, CH₃OH); Anal. Calcd for C₁₇H₂₁N·HCl·0.25H₂O: C, 72.84; H, 8.09; N, 5.00. Found: C, 72.66; H, 8.08; N, 4.99.

5.6. 3β-(4-Chlorophenyl)-2β-ethynyltropane (**9b**)

The procedure for **9a** was followed using 650 mg (1.55 mmol) of **7b** to give 260 mg (65%) of **9b** as a white solid: mp 123–125 °C; ¹H NMR (CDCl₃) δ 7.32–7.18 (m, 4H), 3.40–3.30 (m, 2H), 2.98 (ddd, *J* = 12.9, 5.4, 5.4 Hz, 1H), 2.72–2.65 (m, 1H), 2.33 (s, 3H), 2.30–2.00 (m, 4H), 1.78–1.53 (m, 3H); ¹³C NMR (75 MHz; CDCl₃) δ 141.0, 132.3, 129.7, 128.2, 84.4, 71.9, 66.6, 61.8, 42.3, 41.5, 35.9, 35.1, 26.1, 25.2; MS (ESI) *m/z* 260.4 (M+1). The free base was converted to the hydrochloride salt: mp 145 °C (fusion); [α]_D²⁰ –106.3° (*c* 0.26, CH₃OH); Anal. Calcd for C₁₆H₁₈ClN·HCl·1.25H₂O: C, 60.29; H, 6.80; N, 4.39. Found: C, 60.37; H, 6.89; N, 4.27.

5.7. 3β-(4-Methylphenyl)-2β-(phenylethynyl)tropane (**10a**)

To a stirred mixture of **9a** (40.0 mg, 0.17 mmol), CuI (3.18 mg, 0.017 mmol), and Pd(PPh₃)₄ (6.00 mg, 0.005 mmol) in 1:1 mixture of benzene–TEA (5 mL) at room temperature under nitrogen was added iodobenzene (0.075 mL, 0.67 mmol). The reaction mixture was stirred at 40 °C for 1 h. After cooling to room temperature, the reaction mixture was diluted with EtOAc (50 mL), washed with NH₄Cl (10 mL), brine (3 × 30 mL), and concentrated under reduced pressure. The resultant residue was partitioned between ether (10 mL) and 6 N HCl (10 mL). The aqueous layer was separated, basified to pH 11 with NH₄OH and extracted with EtOAc (3 × 30 mL). The combined EtOAc extracts were washed with brine (3 × 30 mL), dried (Na₂SO₄), and concentrated under reduced pressure. Flash column chromatography on silica gel (12 g Isco prepacked column) using 2% TEA in hexanes afforded **10a** (45.0 mg, 86%) as an oil: ¹H NMR (CDCl₃) δ 7.34–7.08 (m, 9H), 3.47–3.40 (m, 1H), 3.39–3.30 (m, 1H), 3.06 (ddd, *J* = 12.9, 5.4, 5.4 Hz, 1H), 2.93–2.87 (m, 1H), 2.42–2.03 (m, 9H), 1.80–1.54 (m, 3H); ¹³C NMR (75 MHz; CDCl₃) δ 139.8, 136.1, 131.8, 128.8, 128.5, 128.0, 127.3, 124.6, 91.2, 83.9, 66.6, 61.9, 42.5, 42.1, 36.8, 35.3, 26.5, 25.4, 21.2; MS (ESI) *m/z* 316.5 (M+1). The free base was converted to the hydrochloride salt: mp 228 °C (dec); [α]_D²⁰ –229.0° (*c* 0.25, CH₃OH); Anal. Calcd for C₂₃H₂₅N·HCl·0.25H₂O: C, 77.51; H, 7.49; N, 3.93. Found: C, 77.30; H, 7.45; N, 3.92.

5.8. 3β-(4-Chlorophenyl)-2β-(phenylethynyl)tropane (**10b**)

The procedure for **10a** was followed using 78.0 mg (0.30 mmol) of **9b** and 0.13 mL (1.20 mmol) of iodobenzene to give 70.0 mg (70%) of **10b** as an oil: ¹H NMR (CDCl₃) δ 7.29 (s, 5H), 7.20 (m, 4H), 3.50–3.40 (m, 1H), 3.39–3.30 (m, 1H), 3.07 (ddd, *J* = 12.9, 5.4, 5.4 Hz, 1H), 2.92–2.84 (m, 1H), 2.37 (s, 3H), 2.35–2.05 (m, 3H), 1.78–1.53 (m, 3H); ¹³C NMR (75 MHz; CDCl₃) δ 141.5, 132.4, 131.8, 130.0, 128.2, 127.5, 124.2, 90.6, 84.2, 66.5, 61.8, 42.3, 42.0, 36.6, 35.1, 26.5, 25.4; MS (ESI) *m/z* 336.5 (M+1). The free base was converted to the hydrochloride salt: mp 234 °C (dec); [α]_D²⁰ –228.1° (*c* 0.21, CH₃OH); Anal. Calcd for C₂₂H₂₂ClN·HCl·0.5H₂O: C, 69.29; H, 6.34; N, 3.67. Found: C, 69.65; H, 6.12; N, 3.73.

5.9. 3β-(4-Chlorophenyl)-2β-(4-nitrophenylethynyl)tropane (**10c**)

The procedure for **10a** was followed using 78.0 mg (0.30 mmol) of **9b** and 299 mg (1.20 mmol) of 1-iodo-4-nitrobenzene to give 110 mg (96%) of **10c** as a white solid: mp 62–64 °C; ¹H NMR (CDCl₃) δ 8.12–8.02 (m, 2H), 7.40–7.23 (m, 6H), 3.50–3.41 (m, 1H), 3.40–3.30 (m, 1H), 3.12 (ddd, *J* = 12.6, 5.4, 5.4 Hz, 1H), 2.98–2.90 (m, 1H), 2.36 (s, 3H), 2.32–2.05 (m, 3H), 1.80–1.54 (m, 3H); ¹³C NMR (75 MHz; CDCl₃) δ 146.6, 141.1, 132.5, 132.4, 131.3, 129.7, 128.3, 123.4, 97.0, 82.7, 66.3, 61.8, 42.7, 42.1, 36.3, 35.1, 26.3, 25.3; MS (ESI) *m/z* 381.6 (M+1). The free base was converted to the hydrochloride salt: mp 115–117 °C; [α]_D²⁰ –265.3° (*c* 0.23, CH₃OH); Anal. Calcd for C₂₂H₂₁ClN₂O₂·HCl·0.25H₂O: C, 62.64; H, 5.38; N, 6.64. Found: C, 62.44; H, 5.31; N, 6.49.

5.10. 3β-(4-Chlorophenyl)-2β-(4-aminophenylethynyl)tropane (**10d**)

The procedure for **10a** was followed using 78.0 mg (0.30 mmol) of **9b** and 263 mg (1.20 mmol) of 4-iodoaniline to give 65.0 mg (62%) of **10d** as a white solid: mp 188–190 °C; ¹H NMR (CDCl₃) δ 7.27 (s, 4H), 7.01 (d, *J* = 6.0 Hz, 2H), 6.50 (d, *J* = 6.0 Hz, 2H), 3.64 (br s, 2H), 3.45–3.38 (m, 1H), 3.38–3.30 (m, 1H), 3.02 (ddd, *J* = 12.6, 5.4, 5.4 Hz, 1H), 2.88–2.81 (m, 1H), 2.37 (s, 3H), 2.32–2.03 (m, 3H), 1.80–1.53 (m, 3H); ¹³C NMR (75 MHz; CDCl₃) δ 146.0, 141.6, 132.9, 132.2, 130.0, 128.2, 114.7, 113.9, 87.9, 84.5, 66.6, 61.8, 42.2, 42.0, 36.7, 35.1, 26.5, 25.4; MS (ESI) *m/z* 351.3 (M+1). The free base was converted to the hydrochloride salt: mp 240 °C (dec); [α]_D²⁰ –199.2° (*c* 0.26, CH₃OH); Anal. Calcd for C₂₂H₂₃ClN₂·2HCl·1.25 H₂O: C, 59.20; H, 6.21; N, 6.28. Found: C, 59.57; H, 6.11; N, 6.05.

5.11. 3β-(4-Chlorophenyl)-2β-[1-(2-piperidin-1-yl)ethyl]-1H-[1,2,3]triazol-4-yl]tropane (**11**)

To a stirred suspension of **9b** (130 mg, 0.50 mmol) and 1-(2-azidoethyl)-piperidine (77.1 mg, 0.50 mmol) in 1:1 mixture of *t*BuOH–H₂O (4 mL) at room temperature under nitrogen was added a freshly prepared 1 M sodium ascorbate solution (0.50 mL, 0.50 mmol) followed by 0.3 M CuSO₄ solution (0.167 mL, 0.05 mmol). After stir-

ring for 10 h, the reaction mixture was diluted with EtOAc (50 mL), washed with brine (3× 30 mL) and dried (Na₂SO₄). The solvent was concentrated under reduced pressure. Flash column chromatography on silica gel (12 g Isco prepacked column) using 0 → 10% ether in hexanes with 5% TEA afforded **11** (135 mg, 65%) as a white solid: mp 250 °C (dec); ¹H NMR (CDCl₃) δ 7.77 (s, 1H), 7.02–6.92 (m, 2H), 6.83–6.73 (m, 2H), 4.37–4.16 (m, 2H), 3.34–3.09 (m, 4H), 2.60 (t, *J* = 6.0 Hz, 2H), 2.40–2.21 (m, 4H), 2.20–1.92 (m, 6H), 1.80–1.30 (m, 9H); ¹³C NMR (75 MHz; CDCl₃) δ 147.4, 141.2, 131.6, 129.1, 128.0, 124.1, 66.8, 61.8, 58.6, 54.5, 47.7, 45.7, 42.2, 35.6, 35.0, 26.3, 26.2, 25.2, 24.3; MS (EI) *m/z* 414.2 (M⁺). The free base was converted to the hydrochloride salt: mp 145 °C (fusion); [α]_D²⁰ –66.0° (*c* 0.24, CH₃OH); Anal. Calcd for C₂₃H₃₂ClN₅·2HCl·1.5H₂O: C, 53.75; H, 7.26; N, 13.63. Found: C, 54.13; H, 7.21; N, 13.27.

5.12. 3β-(4-Chlorophenyl)-2β-(1-benzyl-1H-[1,2,3]triazol-4-yl)tropane (**12**)

The procedure for **11** was followed using 130 mg (0.50 mmol) of **9b** and 0.066 mL (0.50 mmol) of benzylazide to give 155 mg (79%) of **12** as a white solid: 118–119 °C; ¹H NMR (CDCl₃) δ 7.63 (s, 1H), 7.40–7.27 (m, 3H), 7.10–6.95 (m, 4H), 6.77 (d, *J* = 8.4 Hz, 2H), 5.55 (d, *J* = 15.0 Hz, 1H), 5.31 (d, *J* = 15.0 Hz, 1H), 3.46–3.40 (m, 1H), 3.32–3.18 (m, 3H), 2.30–1.93 (m, 6H), 1.88–1.53 (m, 3H); ¹³C NMR (75 MHz; CDCl₃) δ 148.3, 141.1, 135.9, 131.7, 129.0, 128.9, 128.4, 128.1, 127.3, 123.6, 66.7, 61.8, 53.7, 46.0, 42.2, 35.4, 34.9, 26.3, 25.2; MS (EI) *m/z* 393.2 (M⁺). The free base was converted to the hydrochloride salt: mp 118 °C (fusion); [α]_D²⁰ –110.9° (*c* 0.23, CH₃OH); Anal. Calcd for C₂₃H₂₅ClN₄·HCl·0.75H₂O: C, 62.37; H, 6.26; N, 12.65. Found: C, 62.44; H, 6.29; N, 12.53.

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